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Bis(4-phenylpiperazin-1-yl)methane

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Key indicators

Single-crystal X-ray study T = 298 KMean σ (C–C) = 0.005 Å R factor = 0.060 wR factor = 0.181 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{21}H_{28}N_4$, was synthesized. It is an unusual example of a Z' = 4 structure without obvious pseudosymmetry. The molecules are linked by numerous $C-H\cdots N$ and $C-H\cdots \pi$ hydrogen bonds; there are many $H\cdots H$ short contacts.

Comment

Crystal structures containing several molecules in the asymmetric unit (Z' > 1) include widely distributed species of hydroxy-containing compounds forming strong hydrogenbonded assemblies in crystals (Kuleshova et al., 2003; Lehmler et al., 2002). Approximate symmetry relationships between independent molecules are apparent in many structures with Z' > 1, such as 4-biphenylol (Z' = 2) (Brock & Haller, 1984) and 2-cyclopentyl-2-hydroxy-2-phenylacetic acid (Z' = 2)(Bats et al., 2003). 4-Chloro-2'-biphenyl, a Z' = 4 structure without obvious pseudosymmetry, is an unusual example discovered recently (Lehmler et al., 2002). The title compound, (I), can be regarded as a symmetric molecule; however, the crystal structure has Z' = 4 without obvious pseudosymmetry. The molecules are linked by many C- $H \cdots N$ and $C - H \cdots \pi$ hydrogen bonds; there are many $H \cdots H$ short contacts (Steiner, 2002).



Experimental

The title compound was synthesized by reacting diiodomethane and the sodium salt of 1-phenylpiperazine, which was prepared *in situ* by a modification of the literature method (Garrard & Partridge, 1993), in dry DMF at 303 K for 12 h. Work-up gave the desired product (yield 73.2%), which was purified by recrystallization from hot alcohol, yielding white crystals of (I) (m.p. 397–399 K). ¹H NMR (CDCl₃, 500 MHz): δ 2.69 (*t*, *J* = 4.93 Hz, 8H), 3.04 (*s*, 2H), 3.20 (*t*, *J* = 4.93 Hz, 8H), 6.83–7.28 (*m*, 10H); ¹³C NMR (CDCl₃, 500 MHz): δ 49.1, 51.5, 80.9, 116.0, 119.5, 129.0, 151.4; IR (KBr): ν 2957, 2880, 2821, 1600, 1503, 1452, 1371, 1237, 1147, 1014, 924, 756, 689 cm⁻¹. EI–MS *m*/*z* (%): 337 (*M*⁺, 1.8), 174 (69.8), 120 (100), 77 (47.5).

Crystal data

$C_{21}H_{28}N_4$
$M_r = 336.47$
Monoclinic, $P2_1/c$
a = 19.049(5)Å
b = 9.758 (8) Å
c = 41.950 (10) Å
$\beta = 99.49 \ (2)^{\circ}$
$V = 7691 (7) \text{ Å}^3$
Z = 16

 $D_x = 1.162 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 5814 reflections $\theta = 2.2-20.9^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$ T = 298 (2) KParallelepiped, colourless $0.30 \times 0.20 \text{ mm}$

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Data collection

Siemens SMART CCD area- detector diffractometer	6119 reflections with $I > 2\sigma(I)$ $R_{int} = 0.059$
ω scans	$\theta_{\rm max} = 25.1^{\circ}$
Absorption correction: none	$h = -22 \rightarrow 22$
31805 measured reflections	$k = -11 \rightarrow 11$
13700 independent reflections	$l = -49 \rightarrow 35$
Refinement	
Refinement on F^2	H-atom parameters constrained

Refinement on F^2	H-atom parameters constrain
$R[F^2 > 2\sigma(F^2)] = 0.060$	$w = 1/[\sigma^2 (F_o^2) + (0.0808P)^2]$
$wR(F^2) = 0.181$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.94	$(\Delta/\sigma)_{\rm max} = 0.001$
13700 reflections	$\Delta \rho_{\rm max} = 0.14 \text{ e} \text{ \AA}^{-3}$
902 parameters	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$

2 parameters $\Delta \rho_{\min} = -0.13 \text{ e} \text{ Å}^{-3}$ H atoms were refined isotropically. All H atoms were located by

geometry calculations riding on their parent atoms. Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 1997); software used to prepare material for publication: *SHELXL*97/2 (Sheldrick, 1997).

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Figure 1

Structure of the asymmetric unit of the title compound. Displacement ellipsoids are drawn at the 10% probability level.

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